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# Comparison of n-type Gd<sub>2</sub>O<sub>3</sub> and Gd-doped HfO<sub>2</sub>

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## Abstract

Gd<sub>2</sub>O<sub>3</sub> and Gd-doped HfO<sub>2</sub> films were deposited on p-type silicon substrates in a reducing atmosphere. Gd 4f photoexcitation peaks at roughly 7 and 5 eV below the valence band maximum have been identified using the resonant photoemission of Gd<sub>2</sub>O<sub>3</sub> and Gd-doped HfO<sub>2</sub> films, respectively. In the case of Gd<sub>2</sub>O<sub>3</sub>, strong hybridization with the O 2p band is demonstrated, and there is evidence that the Gd 4f weighted band exhibits dispersion in the bulk band structure. The rectifying (diode-like) properties of Gd-doped HfO<sub>2</sub>-silicon and Gd<sub>2</sub>O<sub>3</sub>-silicon heterojunctions are demonstrated.

## 1. Introduction

The position of the 4f Gd states and the extent of their hybridization is an issue for compound 4f and 5f semiconductors. In the more localized rare earth compounds, Olson *et al* [1] found evidence for 4f delocalization in CeSb<sub>2</sub>, in the form of dispersing Bloch states, while 4f dispersion has been observed in other Ce [2, 3] and Yb [4] compounds. Of course f shell delocalization in metallic systems can also occur, but is more expected for the 5f levels, and Joyce and Durakiewicz, and their colleagues [5–9] and others [10] also found clear evidence of dispersion in the uranium 5f levels in uranium and uranium compounds. In the Gd<sup>3+</sup> and mixed valence systems, Gd 4f hybridization with nearest neighbor atoms is expected [11, 12] and observed. In ErAs(100), strong 4f hybridization is implicated, and a direct confirmation of an occupied 4f band structure might be inferred from the data but could not be established [13, 14].

We have compared the occupied density of states for Gd-doped HfO<sub>2</sub> for various levels of Gd doping, observing changes that could be attributed to either increasing Gd

concentration or the monoclinic to cubic structural phase change that also accompanies increased doping levels [12]. To better assess the Gd 4f interaction with the oxygen nearest neighbors, it is essential to compare the n-type Gd-doped HfO<sub>2</sub> with n-type Gd<sub>2</sub>O<sub>3</sub>, as both are in the monoclinic phase when grown on Si(100) by pulsed laser deposition. Although Gd is expected to be a p-type dopant in HfO<sub>2</sub>, one can fabricate a heterojunction diode of n-type Gd-doped HfO<sub>2</sub> with silicon by overcompensating the Gd acceptor states by donor states introduced by oxygen vacancies at the lower Gd doping levels [12, 15]. This formation of donor states should more easily be accomplished with Gd<sub>2</sub>O<sub>3</sub>.

## 2. Experimental details

The Gd-doped (3 at.%) HfO<sub>2</sub> films were deposited on single crystal silicon (100) p-type (and n-type) substrates using pulsed laser deposition (PLD) at a growth rate of about 0.15 Å s<sup>-1</sup>. (Here, *x*% Gd means the nominal composition Hf<sub>1-x</sub>Gd<sub>x</sub>O<sub>2-0.5x</sub>.) The Gd-HfO<sub>2</sub> target was prepared by standard ceramic techniques using HfO<sub>2</sub> and Gd<sub>2</sub>O<sub>3</sub> powders,

as described elsewhere [15, 16]. The  $\text{Gd}_2\text{O}_3$  was grown on Si(100) using a pure  $\text{Gd}_2\text{O}_3$  target. Before the deposition, the Si(100) substrates were cleaned with diluted HF acid, rinsed with acetone, and then immediately put in vacuum chamber. Before deposition, the surface of the Si wafer was sputter cleaned in a plasma of  $\text{H}_2$  (8%) and Ar (92%) mixture created by a DC sputtering gun operating in the reverse bias mode. The films were deposited at a substrate temperature of  $500^\circ\text{C}$ . The chamber was pumped to a base pressure of  $3 \times 10^{-7}$  Torr and the deposition was carried out in a mixture of  $\text{H}_2$  and Ar (8%  $\text{H}_2$ ) to introduce the necessary oxygen vacancies. The vacuum was maintained at  $10^{-5}$  Torr during the deposition. The doping level was determined from the target composition, with companion measurements using near edge x-ray absorption spectroscopy (NEXAFS), and on separate samples by x-ray emission spectroscopy (XES or EDAX) on similarly prepared samples. The complementary spectroscopies show that the films and the target have essentially the same composition.

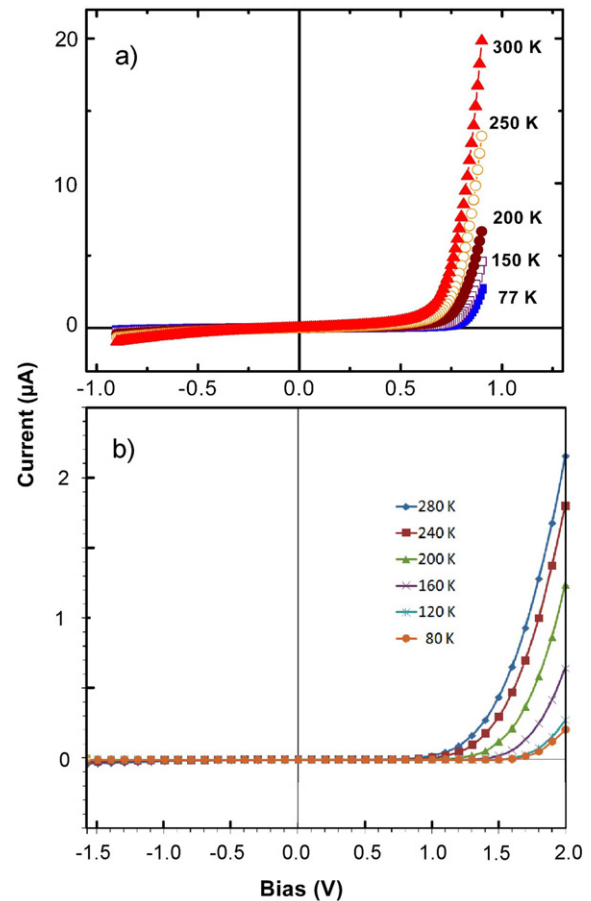
We fabricated several diodes to illustrate the n-type band offset of the  $\text{Gd}_2\text{O}_3$  and 3% Gd-doped  $\text{HfO}_2$  relative to p-type silicon. The heterojunction of  $\text{Gd}_2\text{O}_3$  and Gd-doped  $\text{HfO}_2$  with p-type silicon forms an excellent diode, as shown in figure 1. While these data do not conclusively show the dominant carrier, they suggest that oxygen vacancies can overcompensate the Gd acceptor states in 3% Gd-doped  $\text{HfO}_2$  without completely destroying the semiconductor properties, consistent with the photoemission [12, 15]. On n-type silicon, the heterojunctions do not show clear diode rectification and the  $I(V)$  characteristics are more representative of a resistor, consistent with the picture of oxygen vacancies.

To determine the placement of the Fermi level, angle-resolved photoemission experiments were performed using the 3 m toroidal grating monochromator (3 m TGM) beam line [17] in a UHV chamber previously described [13, 14, 17]. The Fermi level ( $E_F$ ) was established from a gold film in electrical contact with the sample and measurements were carried out at ambient temperatures.

### 3. The structure of the films

Both x-ray diffraction and extended x-ray adsorption fine structure spectroscopy show that 3% Gd-doped  $\text{HfO}_2$  films and  $\text{Gd}_2\text{O}_3$  films grown on Si(100) are highly textured. As noted elsewhere [12, 15], the x-ray diffraction patterns show that the resulting approximately 250 nm thick 3% Gd-doped  $\text{HfO}_2$  films are in a single monoclinic phase with strong texture growth, with about 3% strain compared with the undoped  $\text{HfO}_2$  (figure 2). From the largest peak near  $28^\circ$  ( $2\theta$ ), it is estimated that the lattice spacing for  $\langle 111 \rangle$  is increased by  $d = 0.0030$  nm, from 0.3147(1) nm for the undoped  $\text{HfO}_2$  films to 0.3177(1) nm for Gd-doped samples (figure 2). The peak is shifted to lower angles by  $0.338^\circ$ .

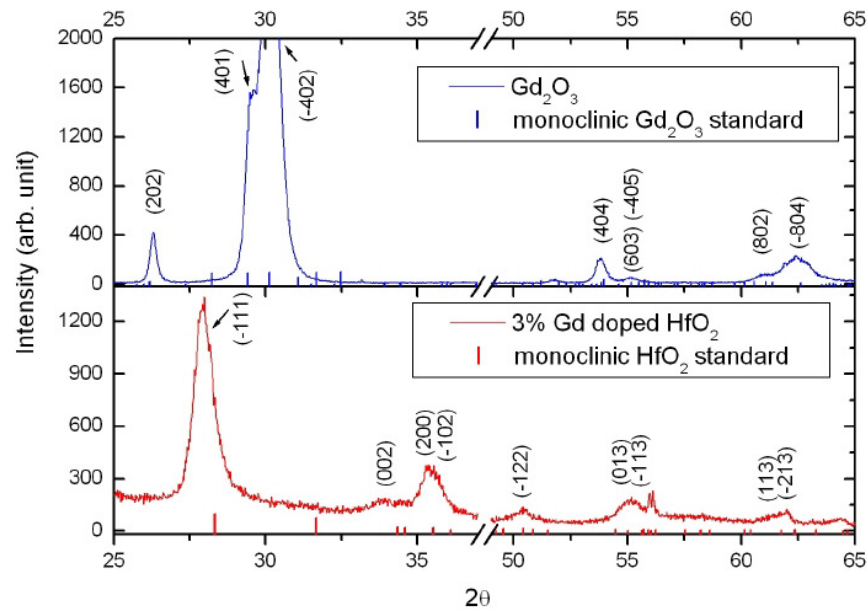
Gd L3 edge x-ray adsorption near edge structure (XANES) and extended x-ray absorption fine structure (EXAFS) spectra of 3% doped sample were collected at the DCM beamline at the Center for Microstructures and Devices (CAMD). Monochromatic light was obtained by using a double crystal monochromator of Lemonnier type [18],



**Figure 1.** Current voltage characteristics for heterojunction diodes constructed from Gd-doped  $\text{HfO}_2$  on p-type silicon (a), and from monoclinic  $\text{Gd}_2\text{O}_3$  on p-type silicon (b), as a function of temperature. These data demonstrate the diode characteristics over a range of temperatures.

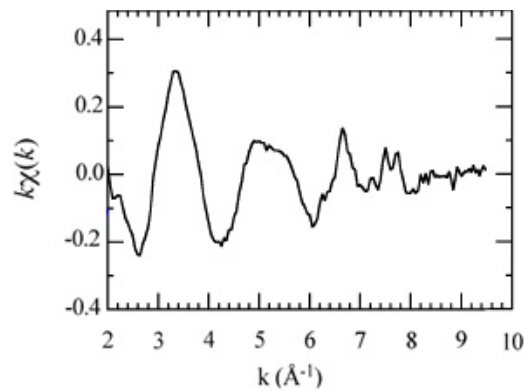
equipped with a Ge(220) crystal pair. The energy resolution was approximately 2 eV and spectra were acquired in the fluorescent yield mode, using a Ge detector (Canberra). Magnitude and imaginary part of the Fourier transformed (FT)  $k\chi(k)$  of the Gd  $L_3$  edge for 3% samples is shown in figure 3. While the magnitude of the Fourier transformed  $k\chi(k)$  is not precisely the pair radial distribution function, this does provide an indication of the radial spacing of atoms in the vicinity of Gd. The first peak at  $\sim 0.8\text{--}2.5$  Å corresponds to the single-scattering contribution of Gd–O pairs. A multi-peak structure at  $\sim 2.5\text{--}5.2$  Å is due to combined contributions of Gd–O, and Gd–Hf single-scattering paths as well as several multiple-scattering paths, as summarized in figure 4. In other words, we find that Gd occupies the Hf site in  $\text{HfO}_2$ , consistent with expectations [12, 15].

For  $\text{Gd}_2\text{O}_3$ , the x-ray diffraction is consistent with highly textured monoclinic  $\text{Gd}_2\text{O}_3$ , not the more expected cubic structure [19, 20]. But the substrate has a strong influence on the texture and crystal structure of the  $\text{Gd}_2\text{O}_3$  film [21, 22], as is clearly the case here. The textured structure is such that the  $\langle 402 \rangle$  planes mostly lie along the surface of the film. Some  $\langle 401 \rangle$  and  $\langle 202 \rangle$  planes are also found to orient parallel to the surface of the film. Only the  $\langle h0l \rangle$  planes (planes parallel to the



**Figure 2.** Part of the XRD pattern of the film is shown in figure. X-ray diffraction patterns of PLD grown  $Gd_2O_3$  (upper panel) and 3% Gd-doped  $HfO_2$  (lower panel). The bar diagrams included in each panel are the standards of monoclinic  $Gd_2O_3$  and  $HfO_2$ , respectively. For 3% Gd-doped  $HfO_2$  (lower panel), the XRD is consistent with that of  $HfO_2$  in a simple monoclinic structure. From the largest peak near  $28^\circ$  ( $2\theta$ ), shown here, we estimated that the lattice spacing for  $(\bar{1}11)$  is increased by  $d = 0.0030(1)$  nm. The peak is shifted to lower angles by  $0.338^\circ$ .

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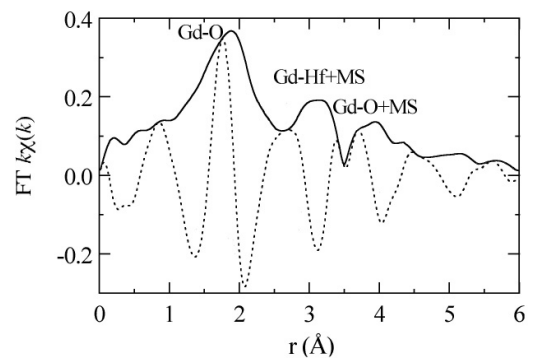


**Figure 3.** The  $k\chi(k)$  of 3% Gd-doped  $HfO_2$  extracted from the Gd  $L_3$ -edge EXAFS spectra.

$b$  axis) are grown parallel to the surface. The lattice spacing along the  $\langle 402 \rangle$  direction is about 0.2965 nm and the lattice spacing for  $\langle 401 \rangle$  and  $\langle 202 \rangle$  is 0.3033 nm and 0.3402 nm, respectively. The  $Gd_2O_3$  unit cell is large with inequivalent Gd (three) and oxygen atoms (in a number of inequivalent sites) as schematically shown in figure 5. The repeat along the surface normal is eight times the  $\langle 402 \rangle$  layer spacing or about 2.372 nm.

#### 4. The occupied density of states

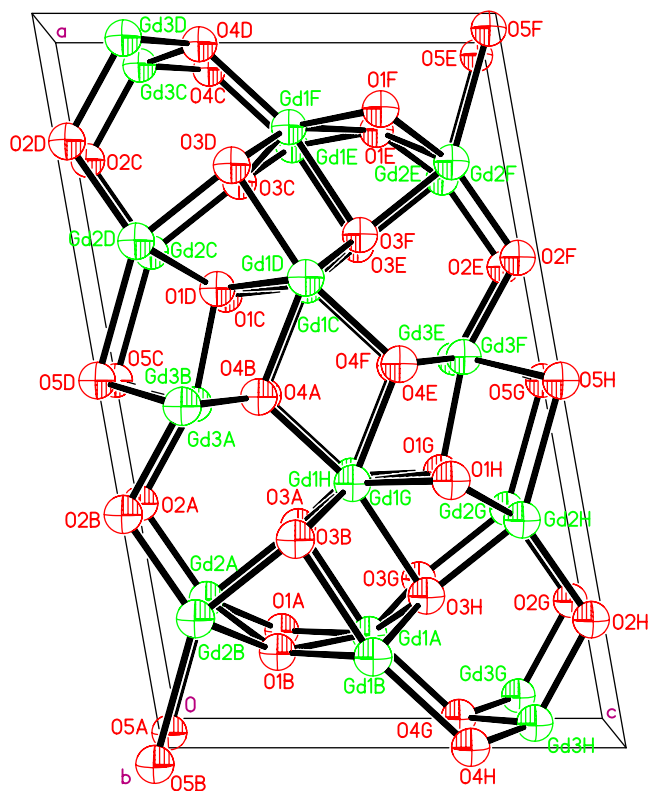
From figure 6, it is clear that the valence band edge is placed well away from the Fermi level for both 3% Gd-doped and undoped  $HfO_2$  films. As reported previously [12, 15], the Hf



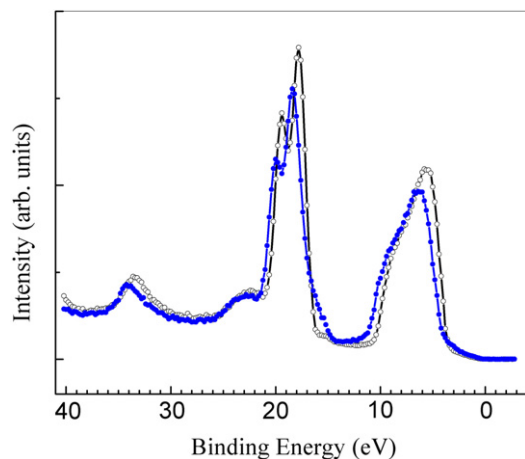
**Figure 4.** The Fourier transform (FT) of 3% Gd-doped  $HfO_2$  EXAFS data. The oscillating curve (dashed lines) is the imaginary part of the Fourier transform of the data. The envelope (solid lines) are the magnitudes of the Fourier transform. The 3% data are transformed with a square window between 2.8 and  $9.4 \text{ \AA}^{-1}$ . The peak assignment is based on  $HfO_2$ .

4f binding energies and valence band edge are similar [23] or slightly larger than those reported elsewhere [24–26]. The shoulder on the broad photoemission peak is at the binding energy of 9–10 eV, which has been already assigned strong Gd 4f weight [12, 15].

The valence band density of states, as determined by photoemission for  $Gd_2O_3$ , are very similar to that observed for 3% Gd-doped films, as indicated in figure 7. The major difference is that for  $Gd_2O_3$ , we require four components at a minimum to fit the valence band spectra, as also indicated in figure 7. This might be expected as from the crystal structure

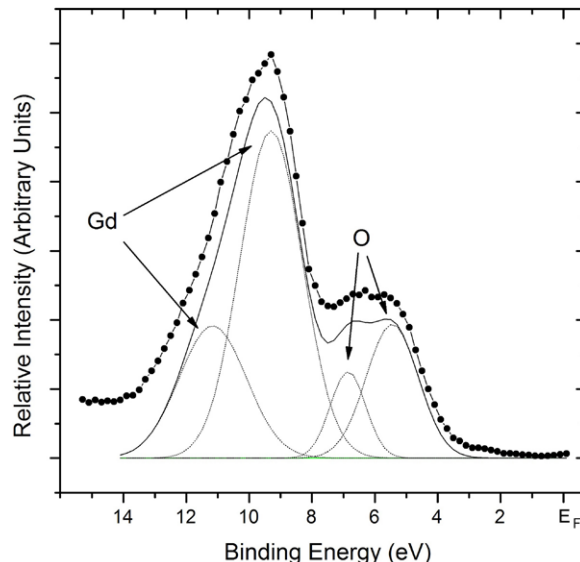


**Figure 5.** Structure of  $Gd_2O_3$ , with the packing of the ions in  $Gd_2O_3$  as viewed along the  $b$ -axis. The green and red spheres represent gadolinium and oxygen atoms respectively. The  $Gd \cdots Gd$  interactions are not shown.



**Figure 6.** Different bands intensities for pristine (open dots) and Gd-doped films of  $HfO_2$  (filled blue dots). The photon energy is 100 eV and the light incidence angle is  $45^\circ$ . All photoelectrons were collected along the surface normal at  $T = 320^\circ C$ .

as there are differences in the crystal field around oxygen and gadolinium: there are two very inequivalent oxygen species, and to a lesser extent true as well of gadolinium, as indicated in figure 5. In fact the gadolinium atoms occupy three inequivalent sites as well, but this does not really lead to a further splitting in the valence band electronic structure sufficient to explain, as discussed below, the two widely separated Gd 4f components at a binding energy of about 8.7–

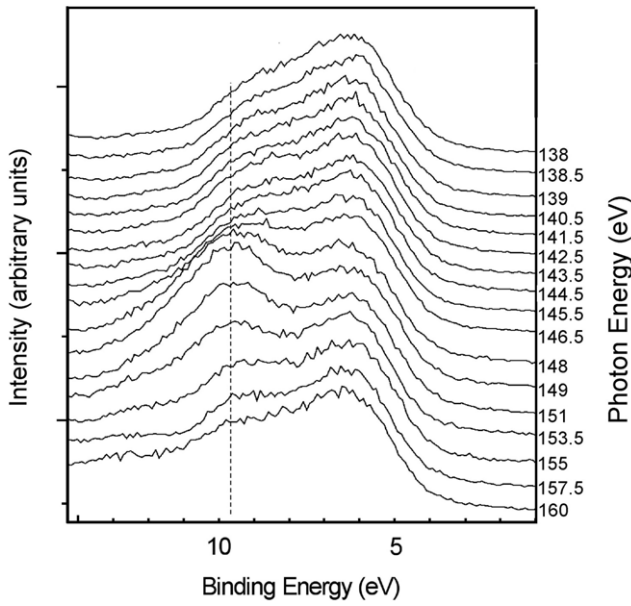


**Figure 7.** The photoemission spectrum of the valence band  $Gd_2O_3(402)$ . The various components contributing to the valence band structure are indicated, and major contributions to the photoemission features indicated. The photon energy is 117 eV and the light incidence angle is  $45^\circ$ . All photoelectrons were collected along the surface normal at  $T = 240^\circ C$ .

9.5 eV below the Fermi level and the ‘shoulder’ at a binding energy of 11–12 eV.

Confirmation that the Gd 4f states contribute to the shoulder at 9–10 eV on the broad photoemission feature from 5 to 10 eV binding energy is found in resonant photoemission (i.e. constant initial state spectroscopy) measurements of Gd-doped  $HfO_2$ . The results are shown in figure 8. The photoelectron intensities from Gd-doped  $HfO_2$ , determined from the feature at about 9.5 eV binding energy (from the Fermi level) is strongly enhanced at about 149 eV photon energy. Similarly for  $Gd_2O_3$ , we see that the components of the valence band at about 8.7–9.5 eV and the ‘shoulder’ at 11–12 eV binding energies are enhanced at about 152 eV, as shown in figure 9 and plotted in figure 10. For comparison, we have plotted this resonant enhancement in the valence band photoemission spectra, for both  $Gd_2O_3$  and Gd-doped  $HfO_2$  films for various photon energies in figures 10(a) and (b) respectively.

It is clear that the resonant enhancements in the photoemission intensity, from this 9.5 eV binding energy final state for Gd-doped  $HfO_2$  and 8.7–9.5 eV for  $Gd_2O_3$ , occur at photon energies corresponding to the core threshold binding energy of the Gd  $4d_{3/2}$  (147 eV) shallow core although at somewhat larger photon energy for  $Gd_2O_3$ . This shift of the photoemission resonance to higher photon energies is expected because of the increase in binding energy of the 4d core and decrease in the conduction band edge binding energy (a placement of the conduction band edge well above the Fermi level [26]). These are in these nominally dielectric oxides, as compared to gadolinium metal. Thus the feature, in the region of 9.5 eV binding energy final state for Gd-doped  $HfO_2$  and 8.7–9.5 eV for  $Gd_2O_3$ , has strong Gd weight. The resonant photoemission process occurs because of an excitation from

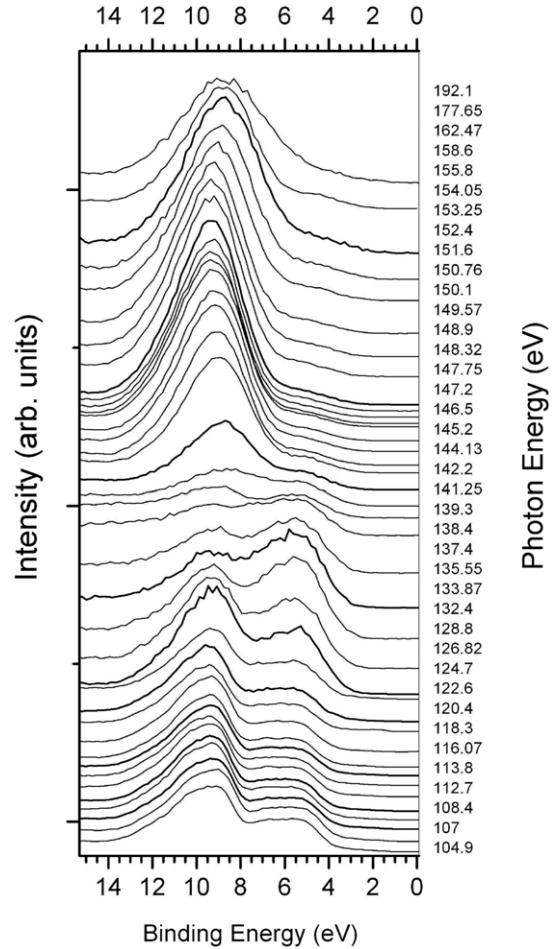


**Figure 8.** Resonant photoemission spectra for photon energies through the  $4d \rightarrow 4f$  resonance for Gd-doped films of  $\text{HfO}_2$ . Light incidence angle is  $45^\circ$ . All photoelectrons were collected along the surface normal.

the  $4d$  cores to a bound state, but with a final state identical to that resulting from direct photoemission from Gd  $4f$  states [27–29]. The photoemission resonance, with changing photon energy, is due to constructive interference between the direct  $5f$  photoionization and a  $4d^{10} 4f^7 \rightarrow 4d^9 4f^8 \rightarrow 4d^{10} 4f^6 + e^-$  super Coster–Kronig transition [28], leading to a classic Fano resonance. The classic Fano resonance shape is very clearly seen for the Gd  $4f$  weighted valence band feature intensities, as plotted in figure 10(a).

These results are generally in agreement with our expectations from the calculated band structure and density of states. The electronic structure of the B-type  $\text{Gd}_2\text{O}_3$  (monoclinic) having six formulae units per monoclinic cell has been calculated using the projected augmented wave method and the generalized gradient approximation (GGA-PBE) with Hubbard  $U$  correction (GGA +  $U$ ) [30] on the Gd  $4f$  energy levels, as implemented in the VASP package [31]. In the GGA +  $U$  calculations, we used  $U = 7.5$  eV and  $J = 0.6$  eV for the Gd  $4f$  orbitals to take the correlation effects into account. The value of the Hubbard  $U$  used here is very close to the value used for the Gd  $4f$  energy levels in the molecular endo-fullerene  $\text{Gd}@C_{60}$  [29]. A similar (although somewhat smaller) value of  $U = 6.7$  eV was found to be appropriate in the examples of the Gd pnictides  $\text{GdX}$  ( $X = \text{N}, \text{P}, \text{As},$  and  $\text{Bi}$ ) compounds [32].

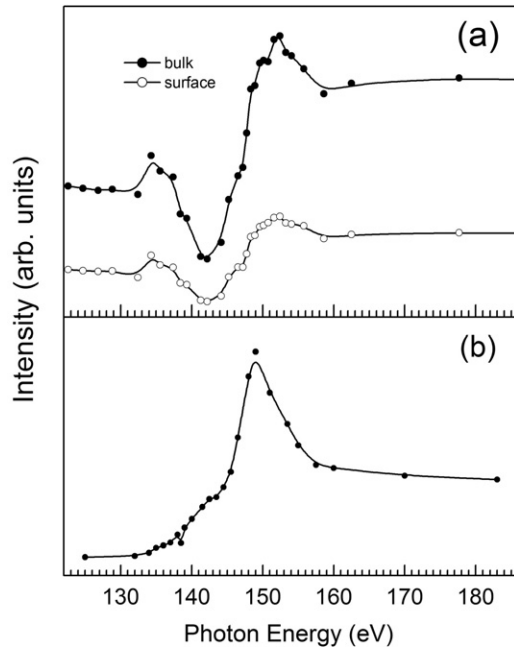
The calculated value of the band gap for the B-type  $\text{Gd}_2\text{O}_3$  (monoclinic) from the GGA +  $U$  method is 3.8 eV. The monoclinic  $\text{Gd}_2\text{O}_3$  density of states (DOS), broadened with a Gaussian width 0.2 eV, is shown in figure 11. The total density of states has been projected onto each atomic species (gadolinium and oxygen) showing the strong Gd  $4f$  character at the DOS peak around  $E = -6.3$  eV and the major oxygen  $2p$  character over the energy range of  $-5$  to



**Figure 9.** The resonant photoemission spectra for photon energies through the  $4d \rightarrow 4f$  resonance for  $\text{Gd}_2\text{O}_3$ . Light incidence angle is  $45^\circ$ . All photoelectrons were collected along the surface normal.

0 eV. This calculation places the occupied Gd  $4f$  levels are a binding energy slightly greater (by about 0.5–1 eV), relative to the valence band maximum, than is observed in experiment, even if only the major Gd  $4f$  component is considered.

There are three different Gd sites Gd(1), Gd(2) and Gd(3) in monoclinic  $\text{Gd}_2\text{O}_3$ , with different coordination numbers and neighboring atoms. To analyze the Gd  $4f$  DOS peak, we calculated the contributions of crystallographically different Gd atoms (figure 5) to the DOS, as indicated by Gd(1), Gd(2), and Gd(3) in the inset of figure 11. While there are clearly differences between Gd(1) and Gd(3), there are some similarities. Each Gd atom site has seven oxygen neighbors, however, for Gd(3), the seventh oxygen neighbor is located farther away than for Gd(1) and Gd(2), so only six neighbor are included in the primary coordination shell. The coordination about Gd(3) is a distorted octahedron with the seventh oxygen atom along a three-fold axis. For Gd(1) and Gd(2), this is not the case, and both have the same  $2mm$  symmetry and seven oxygen neighbors within the primary coordination shell (figure 5). The six oxygen about Gd(1), or Gd(2), form a trigonal prism and the seventh lies along a normal to a prism face. In considering the Gd–Gd distances, Gd(1) and Gd(2) are again similar, having 10 Gd neighbors at distances ranging from 3.28 to 3.98 Å and two more Gd atoms placed further



**Figure 10.** The resonant photoemission intensities through the 4d → 4f resonance for Gd<sub>2</sub>O<sub>3</sub> (a) and Gd-doped films of HfO<sub>2</sub> (b), abstracted from figures 8 and 9. For Gd<sub>2</sub>O<sub>3</sub>, there are seen to be two Gd 4f components at a binding energy of about 8.7–9.5 eV below the Fermi level and the ‘shoulder’ at a binding energy of 11–12 eV, which are assigned as bulk (●) and surface (○) components respectively (see text).

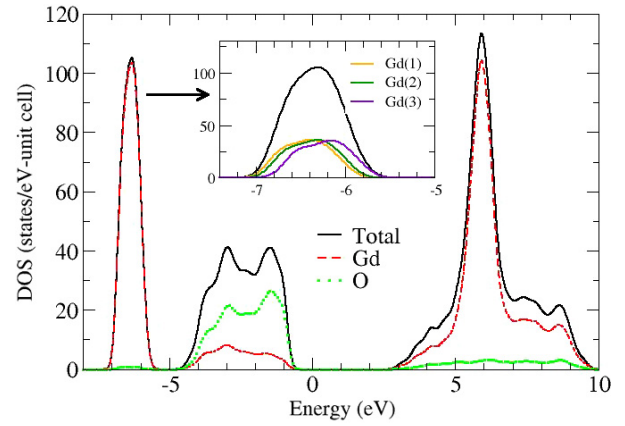
away. On the other hand, Gd(3) has 12 Gd neighbors ranging from 3.58 to 3.87 Å.

As a result, the Gd 4f DOS peak originates from the three distinct types of Gd atoms whose corresponding 4f energy levels are split by as much as ~0.5 eV due to the spin–orbit coupling but are rigidly shifted due to their crystallographic identities. The full linewidth of the Gd 4f DOS peak is about 1.3 eV.

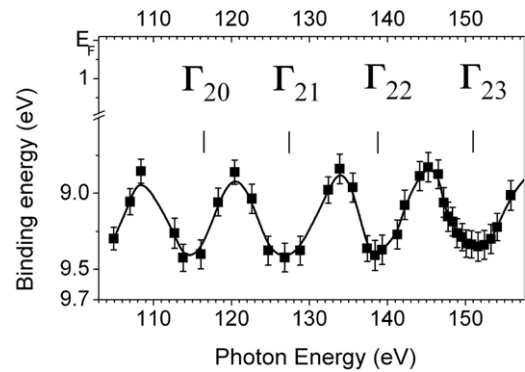
Since the two components we have identified in photoemission of Gd<sub>2</sub>O<sub>3</sub>(402) as heavily Gd 4f, in weight or oscillator strength (on the basis of their photon energy dependence at the 4d<sup>10</sup> 4f<sup>7</sup> → 4d<sup>9</sup> 4f<sup>8</sup> → 4d<sup>10</sup> 4f<sup>6</sup> + e<sup>−</sup> super Coster–Kronig transition, shown in figure 10(a)), are separated by slightly more than 2 eV, we must conclude that the higher binding energy component is, in fact, a consequence of the surface-to-bulk shift. The Gd occupied 4f surface-to-bulk shift is only about 0.4 eV for Gd metal, but is expected to be much larger with oxidation [33, 34], with the surface component at the greater binding energy. Both exhibit very similar resonance. It is the larger intensity component, at 8.7–9.5 eV binding energy, associated with the bulk Gd 4f weighted bands of Gd<sub>2</sub>O<sub>3</sub>, that are of interest in establishing the occupied Gd 4f band structure.

### 5. The 4f band structure of gadolinium oxide

Because of the high degree of order and the strong texture growth along  $\langle 402 \rangle$ , we have been able to measure the bulk band structure along  $\langle 402 \rangle$ . Because the photoelectrons



**Figure 11.** The calculated density of states for monoclinic (type B) Gd<sub>2</sub>O<sub>3</sub>. The monoclinic Gd<sub>2</sub>O<sub>3</sub> density of states (DOS) has been broadened with a Gaussian width 0.2 eV. The total density of states has been projected onto each atomic species (gadolinium and oxygen) showing the strong Gd 4f character at the DOS peak around  $E = -6.3$  eV and the major oxygen 2p character hybridized with Gd 5d orbitals over the energy range of −5 to 0 eV.



**Figure 12.** The dispersion of the Gd 4f component, with changing photon energy. The critical points are indicated, assuming no inner potential and the predicted lattice spacing of 23.7 Å along  $\langle 402 \rangle$ .

are collected along the surface normal in the sequence of photon energy dependent photoemission spectra (figure 9), the binding energy shifts of many of the observed photoemission features is indicative of band dispersion along the electron wave vector normal to the surface,  $k_{\perp}$ . The value of  $k_{\perp}$  can be estimated from the photoelectron kinetic energy making some assumptions about the inner potential  $U_{in}$ :

$$k_{\perp} = \sqrt{\frac{2m}{\hbar^2} \{E_{kin}(\cos(\theta))^2 + U_{in}\}}. \quad (1)$$

The dispersion of the component we attribute to the Gd 4f level at 9–9.5 eV binding energies is summarized in figure 12. The dependence on photon energy shows that the 4f band critical points repetition, whose spacing in wave vector suggests a periodicity of about  $22 \pm 2$  Å perpendicular to the film or along the surface normal, close to the value of 23.72 Å expected from the crystallography.

For a dielectric insulator we would expect that the value of the inner potential to be effectively negligible, as seems to be the case here. We cannot establish that the inner potential to be zero, but from our measurements, that establishes the critical points for the 19th through to the 23rd Brillouin zone, the inner potential is quite small and is no more than 1–2 eV, at most. Assuming a value of zero for the inner potential, and a lattice constant of 23.72 Å, the calculated critical points show very good agreement with the experimental band structure, as indicated in figure 12.

What is clear is that we have strong Gd 4f hybridization with the oxygen in both Gd<sub>2</sub>O<sub>3</sub> and Gd-doped HfO<sub>2</sub> films, and we have a small amount of band dispersion in the band strongly weighted with a Gd 4f contribution for Gd<sub>2</sub>O<sub>3</sub>. The band dispersion is small (about 500 meV), but evident nonetheless, as in the case of Ce [2, 3] and Yb [4] compounds. The Gd 4f weighted bands exhibits band structure yet are placed well away from the Fermi level in the valence band of Gd<sub>2</sub>O<sub>3</sub>.

The band width is about 0.5 eV, which is less than the full Gd 4f linewidth in Gd<sub>2</sub>O<sub>3</sub>, but nonetheless about what is expected from the calculated band width from projected augmented wave method and the generalized gradient approximation (GGA-PBE) with a Hubbard *U* correction (GGA + *U*). What is perhaps surprising is that that we are able to identify this band structure experimentally even though there are some 84 gadolinium bands. This implicates that photoemission selection rules play a role. Such selection rules are indeed likely.

The unoccupied 4f levels of Gd<sub>2</sub>O<sub>3</sub> also have some band width. From the width of the resonant intensities due to the 4d<sup>10</sup> 4f<sup>7</sup> → 4d<sup>9</sup> 4f<sup>8</sup> → 4d<sup>10</sup> 4f<sup>6</sup> + e<sup>-</sup> super Coster–Kronig transition occurring at about 150–152 eV, we can see that the width of this resonant enhancement occurs for a far larger range of photon energies for Gd<sub>2</sub>O<sub>3</sub> than is the case for 3% Gd-doped HfO<sub>2</sub> (figures 10(a) and (b) respectively). Since Gd<sub>2</sub>O<sub>3</sub> and HfO<sub>2</sub> have very similar band gaps (5.4 and 5.7 eV respectively), the pronounced differences in the position of the maximum resonance and the width of the resonance suggest that the unoccupied Gd 4f bands of Gd<sub>2</sub>O<sub>3</sub> have greater band width than is the case for the unoccupied states of Gd in HfO<sub>2</sub>. As was noted for ErAs [13, 14], we expect the unoccupied 4f levels to exhibit dispersion and contribute to the band width of the unoccupied states.

## 6. Conclusion

In summary, we find that superficially there are many similarities in the valence band electronic structure of monoclinic Gd-doped HfO<sub>2</sub> and monoclinic Gd<sub>2</sub>O<sub>3</sub>. We find that with Gd<sub>2</sub>O<sub>3</sub>, there is a surface-to-bulk shift of the Gd 4f levels in Gd<sub>2</sub>O<sub>3</sub>, not seen with Gd-doped HfO<sub>2</sub>, but treating the Gd 4f levels as shallow core levels may not be appropriate. We find that in spite of the high index face adopted by Gd<sub>2</sub>O<sub>3</sub> thin films on Si(100), and the large unit cell of Gd<sub>2</sub>O<sub>3</sub>, Gd 4f band structure is nonetheless and perhaps surprisingly apparent.

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## References

- [1] Olson C G, Chase S J, Canfield P and Lynch D W 1998 *J. Electron Spectrosc. Relat. Phenom.* **93** 175
- [2] Im H J, Ito T, Kim H D, Kimura S, Lee K E, Hong J B, Kwon Y S, Yasui A and Yamagami H 2008 *Phys. Rev. B* **100** 176402
- [3] Andrews A B, Joyce J J, Arko A J and Fisk Z 1996 *Phys. Rev. B* **53** 3317
- [4] Vyalikh D V, Danzenbacher S, Yaresko A N, Holder M, Kucherenko Y, Laubschat C, Krellner C, Hossain Z, Geibel C, Shi M, Patthey L and Molodtsov S L 2008 *Phys. Rev. Lett.* **100** 056402
- [5] Guziewicz E, Durakiewicz T, Butterfield M T, Olson C G, Joyce J J, Arko A J, Sarrao J L, Moore D P and Morales L 2004 *Phys. Rev. B* **69** 045102
- [6] Guziewicz E, Durakiewicz T, Oppeneer P M, Joyce J J, Thompson J D, Olson C G, Butterfield M T, Wojakowski A, Moore D P and Arko A J 2006 *Phys. Rev. B* **73** 155119
- [7] Arko A J, Joyce J J, Andrews A B, Mandrus D, Moshopoulou E, Fisk Z and Canfield P C 1997 *Phil. Mag. B* **75** 603
- [8] Roy L E, Durakiewicz T, Martin R L, Peralta J E, Scuseria G E, Olson C G, Joyce J J and Guziewicz E 2008 *J. Comput. Chem.* **29** 2288
- [9] Guziewicz E, Durakiewicz T, Olson C G, Joyce J J, Butterfield M T, Arko A J, Sarrao J L and Wojakowski A 2006 *Surf. Sci.* **600** 1632
- [10] Molodtsov S L, Boysen J, Richter M, Segovia P, Laubschat C, Gorovikov S A, Ionov A M, Prudnikova G V and Adamchuk V K 1998 *Phys. Rev. B* **57** 13241
- [11] Sabirianov R F, Mei W N, Lu J, Gao Y, Zeng X C, Bolskar R D, Jeppson P, Wu N, Caruso A N and Dowben P A 2007 *J. Phys.: Condens. Matter* **19** 082201
- [12] Losovyj Ya B, Ketsman I, Sokolov A, Belashchenko K D, Dowben P A, Tang J and Wang Z 2007 *Appl. Phys. Lett.* **91** 132908
- [13] Komesu T, Jeong H-K, Choi J, Borca C N, Dowben P A, Petukhov A G, Schultz B D and Palmstrøm C J 2003 *Phys. Rev. B* **67** 035104
- [14] Duan C-G, Komesu T, Jeong H-K, Borca C N, Yin W-G, Liu J, Mei W N, Dowben P A, Petukhov A G, Schultz B D and Palmstrøm C J 2004 *Surf. Rev. Lett.* **11** 531–9
- [15] Ketsman I, Losovyj Ya B, Sokolov A, Tang J, Wang Z, Belashchenko K D and Dowben P A 2007 *Appl. Phys. A* **89** 489–92
- [16] Wang W, Hong Y, Yu M, Rout B, Glass G A and Tang J 2006 *J. Appl. Phys.* **99** 08M117



- [17] Losovyj Ya B, Ketsman I, Morikawa E, Wang Z, Tang J and Dowben P 2007 *Nucl. Instrum. Methods Phys. Res. A* **582** 264–6
- [18] Lemonnier M, Collet O, Depautex C, Esteva J M and Raoux D 1978 *Nucl. Instrum. Methods A* **152** 109
- [19] Haire G and Eyring L 1994 *Handbook on the Physics and Chemistry of Rare Earths* vol 18 (London: Elsevier) p 429
- [20] Horosaki N, Ogata S and Kocer C 2003 *J. Alloys Compounds* **351** 31  
Gila B P, Lee K N, Johnson W, Ren F, Abemathy C R, Pearton S J, Hong M, Kwo J, Mannaerts J P and Anselm K A 2000 *IEEE Trans.* vol IV-1, p 182
- [21] Hong M, Kwo J, Kortan A R, Mannaerts J P and Sergent A M 1999 *Science* **283** 1897
- [22] Yacoby Y, Sowwan M, Stern E, Cross J O, Brewe D, Pindak R, Pitney J, Dufresne E M and Clarke R 2002 *Nat. Mater.* **1** 99
- [23] Komatsu M, Yasuhara R, Takahashi H, Toyoda S, Kumigashira H, Oshima M, Kukuruznyak D and Chikyow T 2006 *Appl. Phys. Lett.* **89** 172107
- [24] Suzer S, Sayan S, Banaszak Holl M M, Garfunkel E, Hussain Z and Hamdan N M 2003 *J. Vac. Sci. Technol. A* **21** 106
- [25] Sayan S, Emge T, Garfunkel E, Zhao X, Wielunski L, Bartynski R A, Vanderbilt D, Suehle J S, Suzer S and Banaszak Holl M M 2004 *J. Appl. Phys.* **96** 7485
- [26] Sayan S, Bartynski R A, Zhao X, Gusev E P, Vanderbilt D, Croft M, Banaszak Holl M M and Suzer S 2004 *Phys. Status Solidi b* **241** 2246
- [27] Dowben P A, Li D, Zhang J and Onellion M 1995 *J. Vac. Sci. Technol. A* **13** 1549
- [28] Kachel T, Rochow R, Gudat W, Jungblut R, Rader O and Cabone C 1992 *Phys. Rev. B* **45** 7276
- [29] Sabirianov R F, Mei W N, Lu J, Gao Y, Zeng X C, Bolskar R D, Jeppson P, Wu N, Caruso A N and Dowben P A 2007 *J. Phys.: Condens. Matter* **19** 082201
- [30] Liechtenstein A I, Anisimov V I and Zaanen J 1995 *Phys. Rev. B* **52** R5467
- [31] Kresse G and Joubert D 1999 *Phys. Rev. B* **59** 1758
- [32] Duan C-G, Sabirianov R F, Liu J, Mei W N, Dowben P A and Hardy J R 2005 *Phys. Rev. Lett.* **94** 237201  
Duan C-G, Sabirianov R F, Mei W N, Dowben P A, Jaswal S S and Tsymbal E Y 2007 *J. Phys.: Condens. Matter* **19** 315220
- [33] McIlroy D N, Waldfried C, Li D, Pearson J, Bader S D, Huang D-J, Johnson P D, Sabirianov R F, Jaswal S S and Dowben P A 1996 *Phys. Rev. Lett.* **76** 2802–5
- [34] Schussler-Langeheine C, Meier R, Ott H, Hu Z, Mazumdar C, Grigoriev A Y, Kaindl G and Weschke E 1999 *Phys. Rev. B* **60** 3449–52